# **COPPER-COIL REACTORS AS A PLATFORM FOR THE** HYDROXYLATION OF ARYL BROMIDES

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Entry	Ligand	Eq. Ligand	Temp (°C)	RT (min)	Conversion (%) <sup>[a]</sup>	Yield (%) <sup>[a]</sup>	Cu (ppm)
1	L1	0.2	150	30	92	88	94 (2%)
2	L4	0.2	150	30	85	82	150
3	L3	0.2	150	30	43	38	24
4	L5	0.2	150	30	28	14	35
5	L6	0.2	150	30	50	15	-
6	L7	0.2	150	30	30	4	-
7	L8	0.2	150	30	38	9	-
8	L2	0.2	150	30	0	0	511
9	-	-	150	30	0	0	26
10	L1	0.1	150	30	64	60	31
11	L1	0.5	150	30	94	87	224
12	11	0.2	170	20	Q1	75	67





### **Gram-scale reaction**



8

12

Hour

16

24



Reaction conditions: aryl bromide (2 mmol), DMF (12 mL), Na<sub>2</sub>CO<sub>3 (aq)</sub> 0.17 M (14 mL). <sup>[a]</sup> Calculated with internal standard.

## Conclusions

- Copper-coil reactors are feasible platforms for the hydroxylation of aryl bromides.
- Copper-coil reactors outperform other copper sources such as Cu(I) and Cu(II) salts in terms of minimum catalyst loading for the reaction and in terms of the reaction rate.
- The reaction has been tested with electron-donor, electron-acceptor aromatic rings and sterically hindered substrates.
- The model has been scaled to the gram scale to prove the robustness of the catalyst over 24 hours.

[1] Org. Process Res. Dev. 2024, 28, 4477–4484. DOI: 10.1021/acs.oprd.4c00402 [2] Synlett 2014, 25(10), 1409-1412. DOI: 10.1055/s-0033-1338634 [3] Synthesis 2019, 51, 251–257. DOI: 10.1055/s-0037-1610398 [4] Adv. Synth. Catal. 2009, 351, 849–854. DOI: 10.1002/adsc.200800758 \* <u>https://pmm.umicore.com</u> (18/03/2025) and <u>https://www.dailymetalprice.com</u> (18/03/2025)

#### **Research funded by:**

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Ministerio de Universidades: FPU 21/05233